



Improving of Winter Cold Hardiness by Glycine Betaine in Strawberry

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ABSTRACT

One of the most important problems of strawberry cultivation in temperate regions is winter cold injuries. Current study investigated impacts of foliar application of glycine betaine (GB) at 0, 0.5, 1, 2 and 4 mM concentrations on the cold hardiness of strawberry. The plants were divided into two groups: one group for evaluation of cold hardiness at temperatures of -6, -9, -12, -15 and -18 °C; and the other for study of some biochemical characteristics. Results showed that GB treatment increased soluble carbohydrate and proline concentrations in both leaf and crown tissues, total protein concentration in leaf, and relative water content in leaf as compared to those in control. Based on LT₅₀ calculated from electrolyte leakage and tetrazolium staining test, the GB application increased cold hardiness in strawberry plant based on its concentration. The highest cold hardiness was found in the 2 mM GB concentration based on electrolyte leakage and tetrazolium staining tests at -13.3 and -15.3 °C. Meanwhile, the lowest values of cold hardiness were observed in the control treatments based on electrolyte leakage and tetrazolium staining tests at -10.2 and -11.0 °C. Significant correlations were found between soluble carbohydrate and proline concentrations in leaf and crown, and total protein concentration with LT₅₀ calculated from electrolyte leakage and tetrazolium staining test. We conclude that application of 2 mM GB has the capacity to increase the freezing tolerance of strawberry and could be used as a prophylactic tool to reduce winter cold injury.

Introduction

Strawberry (*Fragaria* × *ananassa* Duch.), is one of the most important small fruits cultivated in many temperate regions of the world (Hancock, 1999). One of the main difficulties of strawberry cultivation in temperate regions is sporadic chilling injury at winter (Karami et al., 2018), which leads to

huge economic loss during production season (Zhang et al., 2019; Hancock, 1999). Recently, due to global warming and instability of snow cover in the temperate regions, strawberry vulnerability to winter cold has increased (Lukoseviciute, 2014). Although hardiness of different cultivars varied, it was generally reported that, in unprotected conditions, the strawberry crowns were severely damaged below -9 °C; while acclimated plants injured

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below -12 to -15 °C (Galletta and Himmelrick, 1990; Nestby and Bjorgum, 1999; Maughan et al., 2015).

Cold tolerance in plants is a complex process that is influenced by several factors such as temperature, day length, plant maturity, tissues water content, nutritional status, physiological age, and dormancy progress (Hinch and Zuther, 2014; Ouellet and Charron, 2013). There are various strategies to deal with cold stress including selection of cold tolerant genotypes, breeding of new resistance genotypes, site selection, and use of proper agricultural practices (Sarikhani et al., 2014; Iba, 2002). In addition, practice of chemicals and plant growth regulators could also increase cold hardiness (Sarikhani et al., 2014; Khandan-Mirkohi et al., 2017; Nasibi et al., 2020). There have been reports of successful use of glycine betaine (GB) in combating environmental stresses such as water, salinity, and cold stresses in plants (Demiral and Türkan, 2004; Chen and Murata, 2008; Aras and Esitken, 2013; Adak, 2019). It is one of the four compounds of ammonium that is synthesized in response to stresses in many plant species (Ashraf and Foolad, 2007). Exogenous use of GB in plants that lack or contain very little of it facilitates response to adverse effects of environmental stress (Yang and Lu, 2005).

GB levels in the leaves of strawberry plants increased during cold treatment, and reported to be effective in inducing cold tolerance in response to ABA and natural cold in strawberry (Rajashekar et al., 1999). Due to the increased concentration of GB in stress conditions, it seems that increasing its concentration through foliar application also increases the resistance to stress in plants. Various studies have been conducted on the use of GB for cold tolerance in plants. Application of GB during growing season in maize (Chen et al., 2000), tomato (Park et al., 2006) and strawberry (Aras and Esitken, 2013) decreased chilling injury. Aras and Esitken (2013) reported the survival of most

strawberry bushes at -2.3 °C during the growing season after GB foliar application. Given the positive effects of GB on reducing the adverse effects of cold during the growing season on many plants and the threat of winter cold problem, studies targeting cold tolerance are of vital importance for developing strawberry cultivation in cold regions.

In this study, the effect of application of GB on winter cold hardiness of strawberry cv. Paros, as a short day cultivar and suitable for field conditions, was investigated. Paros is one of the popular strawberry cultivars, and as a semi-tolerant cultivar to winter cold injuries, it is widely planted under open-field condition in temperate regions.

Materials and Methods

Plant material and experiment site

The study was carried out in the research center of Department of Horticultural Science, Bu-Ali Sina University, Hamedan, Iran in 2018 and 2019. In late September, the daughter plants of strawberry cv. Paros were cultivated in five-liter plastic pots containing an equal proportion of crop soil, sand, and decayed manure.

Application of GB

To prepare GB solutions, the required amount of GB (MW = 153.6) was dissolved in distilled water and brought to the desired volume. GB at concentration of 0.5, 1.0, 2.0, and 4.0 mM was sprayed twice on the plants foliage containing 0.05% v/v Tween-20. The first stage of treatment was done on 5th of October and the next round was repeated on 20th of October. For the control plants, distilled water containing Tween-20 was used. After application of treatment, the plants were divided into two groups: one group for evaluation of biochemical traits and relative water content, and the other for evaluation of cold hardiness with simulated cold. Biochemical traits were obtained 7- 10 days after the second treatment of GB during which

the plants were kept under field condition (Fig. 1). The cold-hardiness of canes was evaluated using a programmable freezing room. The starting temperature was 4 °C, and the freezer was programmed to reduce the temperature gradually by 3 °C h⁻¹ to -6 °C, -9

°C, -12 °C, -15 °C, -18 °C, then hold each temperature for 3 h. At the end of each temperature treatment, pots were taken out and the injury rates were evaluated using electrolyte leakage measurement and the tetrazolium staining test.

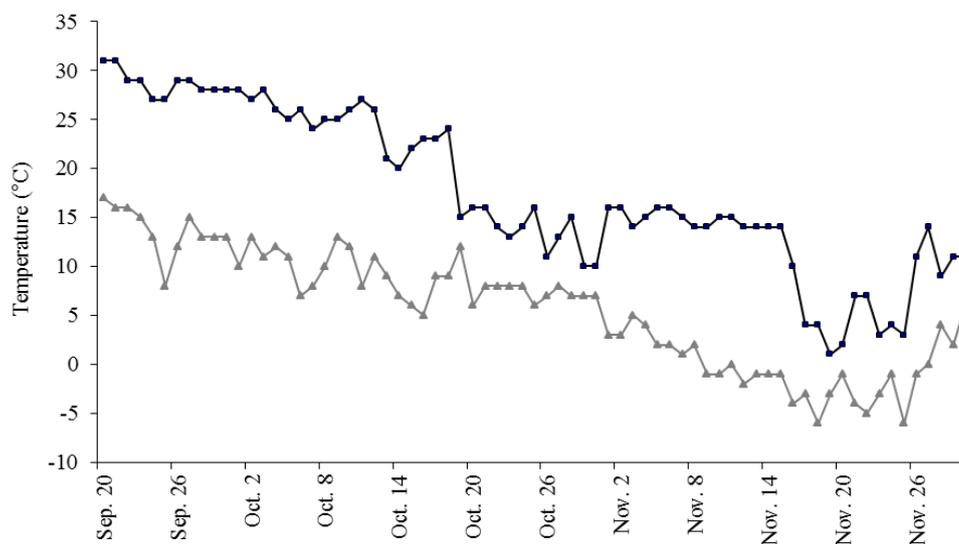


Fig. 1. Daily maximum (■) and minimum (▲) air temperatures at the research site during the experiment.

Chlorophyll concentration

The Porra et al. (1989) method was used to measure chlorophyll concentration in the leaf samples. Briefly, 0.2 g of fresh fully developed leaf tissue was homogenized in 5 mL of 80% acetone. The solution was centrifuged (Herolab model, Germany) for 5 min at 4000 rpm. After the supernatant was removed, the above steps were repeated twice with 80% acetone. The total supernatant volume brought to 20 mL. The absorbance of the samples was then measured by spectrophotometer (UV-1280, Shimadzu, Japan) at 645 and 664 nm. Chlorophyll concentration was calculated based on mg/g FW using the following equation:

$$\text{Total chlorophyll} = (17.76 A_{645}) + (7.34 A_{664})$$

Total soluble carbohydrate concentration

To measure the soluble carbohydrate content, the Paquin and Lechasseur (1979) method was used. For this purpose, 0.2 g of leaf tissue was

carefully grounded using 5 mL of 95% ethanol in a mortar, and then the supernatant was separated. Later 5 mL of 70% ethanol was added to the previous residues. This procedure was repeated twice and the extract was centrifuged for 10 min at 4000 rpm. Next, 0.1 mL of the extract was mixed with 3 mL of freshly prepared anthrone (150 mg anthrone + 100 mL of 72% sulfuric acid). This solution was incubated in a hot water bath at 95 °C for 10 min, and its absorbance was read using a spectrophotometer (UV-1280, Shimadzu, Japan) at 625 nm. Finally, the concentration of soluble carbohydrate was determined by comparing with the standard curve of glucose based on mg/g FW.

Proline concentration

The Bates et al. (1973) method was used to measure proline concentration. First, 0.2 g of leaf tissue was grounded in liquid nitrogen in mortar, and then 10 mL of 3% sulfosalicylic acid (3 g of sulfosalicylic acid in 100 mL

distilled water) was added. The extract was later centrifuged at 4000 rpm for 10 min at 4 °C. Next, 2 mL of the supernatant was removed and poured into the test tube, giving it 2 mL of glacial acetic acid and 2 mL of freshly prepared ninhydrin (1.25 g ninhydrin + 30 mL of glacial acetic acid + 20 mL of 6 M phosphoric acid). Then, the samples were incubated in a hot water bath at 80 °C for one hour. After that, the samples were kept on ice for 5 min, and then 4 mL of toluene was added to each sample before being shaken vigorously for 30 seconds. Finally, the absorbance of above phase was read by a spectrophotometer (UV-1280, Shimadzu, Japan) at 520 nm. To calculate the concentration, the absorbance of the samples was compared with the standard curve, and the final concentration of proline was calculated in mg/g FW.

Concentration of leaf soluble proteins

The Bradford (1976) method was used to measure soluble proteins in leaf sample. Briefly, 0.3 g of frozen leaf tissue was mixed with 6.25 mL of extraction buffer (for one liter, 121.14 g Trice dissolved in water and pH adjust 6.8 using 0.1 N HCl) and placed at 4 °C for 24 h. Then, the samples were grounded again in a mortar in buffer solution and centrifuged at 6,000 rpm for 20 min. Later, 100 µL of the supernatant was carefully removed and 5 mL of fresh Biuret reagent was added. To prepare the Biuret reagent, 100 mg of Coomassie Brilliant Blue G-250 was mixed with 50 mL of ethanol and brought to 800 mL volume with distilled water. Next, 100 mL of 85% phosphoric acid was added to it and final volume was reached 1000 mL with distilled water. Finally, the absorbance of the solution was read using a spectrophotometer (UV-1280, Shimadzu, Japan) at 595 nm, and the protein concentration was determined by bovine albumin standard curve.

Leaf relative water content (RWC)

This trait was measured according to Arndt et al., (2015) method. For this purpose, one square

cm leaf pieces were selected from its healthy area (without the main veins) and weighed. Then, fragments were placed in a petri dish containing distilled water for 24 hours and stored at 4 °C in the refrigerator away from the illumination to allow for cell turgidity. They were later removed from distilled water, and let excess water drained using filter paper before being weighed. Next, the samples were placed in an oven at 72 °C for 48 hours and dry weights of specimens were measured. Finally, the RWC was calculated using the following equation:

$$\text{RWC (\%)} = [(\text{FW}-\text{DW}) / (\text{TW}-\text{DW})] * 100$$

Where FW = fresh weight, DW = dry weight and TW = turgid weight.

Electrolyte leakage

Electrolyte leakage of leaf and crown samples was performed according to the method of Bajji et al. (2002). According to this method, after washing the samples with distilled water, fresh and fully developed leaves/crowns were cut into one cm² pieces and placed in 25 mL distilled water. The specimens were then placed on a shaker at 100 rpm for 2 h, followed by storage at room temperature and dark for 24 h. Later, the primary electrolyte conductivity (EC₁) was read using an EC-meter (model 720, Inolab, Germany). Next, the samples were autoclaved at 120 °C for 20 min, cooled, and secondary electrolyte conductivity (EC₂) was measured. Finally, the electrolyte leakage of the samples was calculated using the following equation:

$$\text{Electrolyte leakage (\%)} = \text{EC}_1 * 100 / \text{EC}_2$$

The lethal temperature (LT) at which 50% of total electrolyte leakage (EL) occurred in leaf and crown tissues (EL-LT₅₀) was calculated based on the method described by Andrews et al. (1984).

Tetrazolium staining test

First, 1% tetrazolium solution was prepared by dissolving 1 g of 2,3,5-triphenyltetrazolium

chloride in 100 mL distilled water. Cold temperature treated samples were analyzed for survival using tetrazolium staining test. The crown samples were immersed in the tetrazolium solution and kept at room temperature and dark for 24 h. Damage was assessed using Leica binocular (MS5, Heerbrugg, Switzerland) and based on the amount of tetrazolium staining of the tissues using a binominal scale as the followings: 1 = no damage, 2 = 100% phloem damage. The specimens that were red in the skin and phloem were considered alive and those lacking red in this area were considered as dead tissue, and then the damage was calculated using the following equation:

$$\text{Damage rate (\%)} = \frac{\text{damaged sample}}{\text{total no. of samples}} \times 100$$

The lethal temperature at which 50% of the crown tissues died estimated by tetrazolium staining test (T-LT₅₀) was calculated by fitting response curves.

Data analysis and experimental design

Data of electrolyte leakage and tetrazolium staining test were analyzed via a factorial experiment in completely randomized design with two factors including GB concentrations and low temperature treatment. In addition, data

of T-LT₅₀ and EL- LT₅₀ was evaluated in a completely randomized design. The biochemical properties including total chlorophyll concentration, leaf soluble carbohydrate concentration, proline concentration of leaf and crown, leaf protein concentration and relative water content were also analyzed in a completely randomized design. Data were analyzed using SAS software (Version 9.1; SAS Institute, 2003) and the means were compared using Duncan's multiple range tests at the 5% level of significance. Correlation coefficients between biochemical properties and T-LT₅₀ and EL- LT₅₀ values were calculated using Pearson's correlation coefficient.

Results

Total chlorophyll concentration

According to the analysis of variance, the effect of GB treatment on total chlorophyll concentration was significant at 5% level (Table 1). The highest concentration of total chlorophyll was observed in 2 mM GB treated plants (3.12 mg/g FW). Also, the lowest total chlorophyll concentration was detected in 4 mM GB treatment with no significant difference with control and 0.5 and 1 mM GB treatments (Table 2).

Table 1. Variance analysis of glycine betaine effect on biochemical properties and RWC of strawberry

| Sources of variance | DF | Mean of square | | | | | Leaf RWC |
|---------------------|----|-------------------|---------------------------|--------------|---------------|--------------|----------|
| | | Total chlorophyll | Leaf soluble carbohydrate | Leaf proline | Crown proline | Leaf protein | |
| GB concentration | 4 | 0.506* | 95.414** | 4.940** | 10.211** | 1.591** | 7.676** |
| Error | 10 | 0.098 | 3.682 | 0.189 | 0.451 | 0.074 | 1.383 |
| CV (%) | - | 12.99 | 7.01 | 13.51 | 16.52 | 8.13 | 1.26 |

** and * significant at 1% and 5%, respectively.

Table 2. Effect of glycine betaine on biochemical properties and RWC of strawberry

| Glycine betaine concentration (mM) | Total chlorophyll (mg/g FW) | Leaf soluble carbohydrate (mg/g FW) | Leaf proline (mg/g FW) | Crown proline (mg/g FW) | Leaf soluble protein (mg/g FW) | Leaf RWC (%) |
|------------------------------------|-----------------------------|-------------------------------------|------------------------|-------------------------|--------------------------------|--------------|
| 0 (control) | 2.18b | 21.73c | 1.38c | 1.59b | 2.58c | 94.8a |
| 0.5 | 2.01b | 23.54c | 2.43b | 2.63b | 3.13b | 94.7a |
| 1 | 2.28b | 31.27b | 4.06a | 5.19a | 3.37b | 93.7b |
| 2 | 2.81a | 35.14a | 4.50a | 5.81a | 4.56a | 92.1c |
| 4 | 1.98b | 25.12c | 3.72a | 5.12a | 3.18b | 91.1d |

In each column, means with the same letters are not significantly different at $P \leq 0.05$ by Duncan's multiple range test.

Leaf soluble carbohydrate concentration

According to the analysis of variance, the effect of GB treatment on leaf carbohydrate concentration was significant at 1% level (Table 1). The highest amount was observed in 2 mM GB (35.14 mg/g FW), and then in 1 mM GB (31.27 mg/g FW). The lowest leaf carbohydrate content was observed in control and there was no significant difference among control, 0.5 mM and 4 mM GB concentrations (Table 2).

Proline concentration of leaf and crown

The effect of GB treatment on leaf and crown proline concentration was significant at 1% level (Table 1). In both leaf and crown tissue, the highest proline concentration was observed in 2 mM GB treatment which showed no significant differences with 1 mM and 4 mM GB concentrations. The lowest leaf proline content was observed in control plants. The lowest amount of crown proline was observed in control plants with no significant difference from 0.5 mM GB (Table 2).

Leaf protein concentration

The effect of GB treatment on leaf protein concentration was significant at 1% level (Table 1). The highest amount was observed in 2 mM GB (4.55 mg/g FW) followed by 1, 4 and 0.5 mM GB, respectively, which showed no significant differences among them. The lowest leaf protein was also detected in control treatment (Table 2).

Relative content of leaf water

The effect of GB treatment on leaf RWC was significant at 5% level (Table 1). The highest amount of RWC was observed in control treatment (94.8%), which had no significant difference with 0.5 mM GB treatment. By increasing GB concentration from 1 mM to 4 mM, RWC reduced significantly (Table 2).

Electrolytic leakage of leaves and crowns

The effect of temperature, GB and their interaction on electrolytic leakage of leaf and crown were significant at 1% level (Table 3).

In both leaf and crown samples, the lowest electrolytic leakage was observed in 2 mM GB treated plants under -6 °C; while the highest one was seen in control plants treated at -18 °C. In leaf sample, at -6 °C, with increasing GB concentration up to 2 mM, the electrolytic leakage decreased. Although no significant difference was observed at -9 °C between 1 and 2 mM GB treatments, at -12 °C the lowest electrolytic leakage was observed at 2 and 4 mM GB treatments. Moreover, at -15 °C the lowest electrolytic leakage was observed at 2 mM GB treatment (Fig. 2a). In crown samples at all treated temperatures, by increasing GB concentration up to 2 mM, electrolytic leakage decreased, while it increased subsequently at 4 mM GB (Fig. 2b).

The effect of GB treatment on leaf and crown EL-LT₅₀ was significant at 1% level (Table 4). In both leaf and crown, the lowest EL-LT₅₀ was observed in GB at 2 mM and then in 1, 4 and 0.5 mM GB, which showed significant differences. The highest EL-LT₅₀ values were observed for leaf and crown tissues in control treatment (Table 5).

Crown injuries estimated from tetrazolium staining test

Analysis of variance revealed that the effect of temperature and GB, and their interaction on crown injury, estimated by tetrazolium test, were significant at the 1% level (Table 3). The highest percentage of crown survival was observed in the 2 mM GB treated plants under low temperature of -6 °C. However, the lowest percentage of crown survival was observed in control plants under -18 °C. In all treated temperatures, by increasing GB concentration up to 2 mM, crown survival rate was increased and then decreased in 4 mM GB treated plants (Fig. 2c).

According to the analysis of variance, the effect of GB treatment on T-LT₅₀ value of crown was significant at 1% level (Table 4). Means comparisons showed the lowest T-LT₅₀ value were: 2 mM GB followed by 1, 4 and 0.5

mM GB treated plants, with significant differences with each other. The highest T-LT₅₀ value was observed in the control (Table 5).

Results showed a significant correlation between EL-LT₅₀ and chlorophyll concentration at 5% level. Also, there were significant correlations between leaf EL-LT₅₀ with leaf soluble carbohydrate, leaf proline, crown proline, leaf protein and relative water content. Also, significant correlations were

found between EL-LT₅₀ of crown with leaf soluble carbohydrate, leaf proline, crown proline, leaf protein, relative water content and leaf EL-LT₅₀. The results showed a significant correlation between T-LT₅₀ and RWC at 5% level, as well. Meanwhile, significant correlations were found between T-LT₅₀ and EL-LT₅₀ of leaf sample with EL-LT₅₀ of crown samples (Table 6).

Table 3. Variance analysis for the effects of glycine betaine and low temperature on strawberry leaf and crown electrolytic leakage and crown tetrazolium test

| Sources of variance | DF | Mean of square | | |
|--------------------------------|----|---------------------------|----------------------------|------------------------|
| | | Leaf electrolytic leakage | Crown Electrolytic leakage | Crown tetrazolium test |
| GB concentration | 4 | 1185.235** | 1089.570** | 2727.833** |
| Temperature | 4 | 55.908** | 59.335** | 778.666** |
| temperature × GB concentration | 16 | 34.345** | 16.582** | 113.75** |
| Error | 50 | 0.446 | 1.679 | 10.333 |
| CV (%) | - | 1.42 | 2.54 | 6.44 |

** significant at 1%.

Table 4. Variance analysis of glycine betaine effect on leaf and crown EL-LT₅₀ and T-LT₅₀ of strawberry

| Sources of variance | DF | Mean of square | | |
|---------------------|----|--------------------------|---------------------------|--------------------|
| | | Leaf EL-LT ₅₀ | Crown EL-LT ₅₀ | T-LT ₅₀ |
| GB concentration | 4 | 27.196** | 22.346** | 22.334** |
| Error | 10 | 0.115 | 0.144 | 0.109 |
| CV (%) | - | 3.13 | 4.06 | 2.73 |

** significant at 1%

Table 5. Effect of glycine betaine effect on leaf and crown EL-LT₅₀ and T-LT₅₀ of strawberry

| Glycine betaine concentration (mM) | Leaf EL-LT ₅₀ | Crown EL-LT ₅₀ | T-LT ₅₀ |
|------------------------------------|--------------------------|---------------------------|--------------------|
| 0 (control) | -9.6e | -10.2e | -11.0e |
| 0.5 | -10.3d | -10.8d | -12.8d |
| 1 | -13.2b | -12.1b | -14.2b |
| 2 | -15.8a | -13.3a | -15.3a |
| 4 | -11.3c | -11.4c | -13.6c |

In each column, means with the same letters are not significantly different at $P \leq 0.05$ by Duncan's multiple range test.

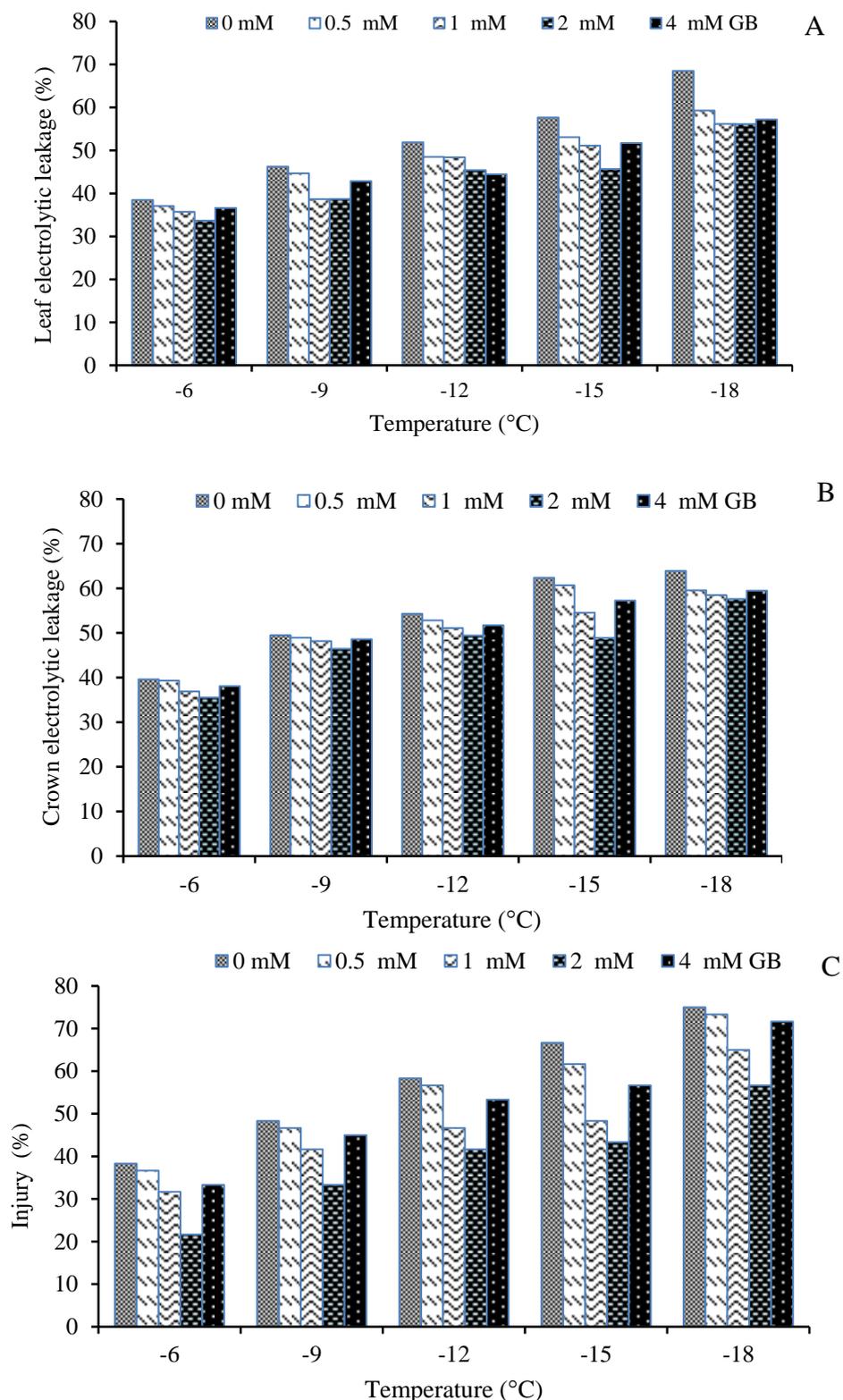


Fig. 2. Glycine betaine (GB) effect on leaf (A) and crown (B) electrolyte leakage and crown injuries (C) based on tetrazolium staining test under different low temperature in strawberry. In each part, mean followed by the same lower letter are not significantly different at $P \leq 0.05$ by Duncan's multiple range test.



Fig. 3. Comparison of the viability of strawberry crown using tetrazolium staining test after cold treatment; Left: Crown staining belonged to the 2 mM glycine betaine without remarkable damage at $-12\text{ }^{\circ}\text{C}$ cold treatment, Right: the crown in this treatment showed defective staining indicating injuries.

Table 6. Pearson correlation coefficients between EL-LT₅₀ and T-LT₅₀, and soluble carbohydrate, total protein, and proline concentrations and relative water content in strawberry following glycine betaine treatment

| | Total chlorophyll | Soluble carbohydrate concentration | Leaf proline concentration | Crown proline concentration | Total protein concentration | RWC | Leaf EL-LT ₅₀ | Crown EL-LT ₅₀ |
|---------------------------|-------------------|------------------------------------|----------------------------|-----------------------------|-----------------------------|----------|--------------------------|---------------------------|
| Leaf EL-LT ₅₀ | -0.563* | -0.933** | -0.916** | -0.855** | -0.790** | +0.657** | 1 | - |
| Crown EL-LT ₅₀ | -0.582* | -0.927** | -0.889** | -0.861** | 0.866** | +0.675** | +0.974** | 1 |
| Crown T-LT ₅₀ | -0.673** | -0.952** | -0.816** | -0.817** | 0.899** | +0.628* | +0.962** | +0.972** |

**,* significant at $P \leq 0.01$ and $P \leq 0.05$, respectively.

T, tetrazolium stain test; EL, electrolyte leakage test.

Discussion

In the present study, the effect of GB on biochemical traits was significant. Increase in chlorophyll concentration after some GB treatments could occur for two reasons: increasing chlorophyll synthesis and preventing chlorophyll degradation. Exogenous GB treatment in strawberry (Aras and Esitken, 2013) and grapevine (Mickelbart et al., 2006) increased total chlorophyll concentration. Reduction of low temperature-degraded chlorophyll due to increased chlorophyllase activity (Meng et al., 2009) is another reason for the higher chlorophyll concentration in GB treated plants (Yildirim et al., 2015). In the present study, only the 2 mM GB increased total chlorophyll concentration and no significant difference was observed between the other treatments and those of control. Although high GB concentration increased chlorophyll content (Mickelbart et

al., 2006), while no effect of 4 mM GB is odd. Moreover, low concentrations of GB had also no effect on increasing total chlorophyll.

There have been several reports of an increase in soluble carbohydrates in plants treated with GB (Ali and Ashraf, 2011; Weibing and Rajashekar, 2018). One of the main reasons for such increase seems to be related to the plant's need for carbohydrate intake for amino acid synthesis. Since GB contains nitrogen and its biosynthesis requires NADPH, it is costly for plants to consume soluble carbohydrates (Gorham, 1995). In this study, the concentration of soluble carbohydrate increased with rising concentration of GB from 0 to 2 mM but decreased in 4 mM treatment. Previous research reported increasing carbohydrate content under GB treatment in beans (Weibing and Rajashekar, 2018), tomato (Mickelbart et al., 2006) and maize (Ali and Ashraf, 2011).

The production and accumulation of osmotic preservatives occur in a wide range of plant species in response to environmental stresses such as drought, salinity, and cold (Fariduddin et al., 2011). Exogenous application of GB increases the accumulation of these molecules under stress conditions. Proline, as an amino acid, naturally accumulates in higher plants under stress conditions (Kavi Kishore et al., 2015; Hirt and Shinozaki, 2004). Proline is an osmotic potential regulator under stress conditions and stabilizes cellular structures (including cell membranes and proteins), eliminating free radicals, and neutralizing cellular redox potential (Irigoyen et al., 2010). GB synthesizes osmotic regulators such as proline and thereby induces resistance to stress (Chaum et al., 2006). In the present study, the proline content of leaf and crown elevated by increasing GB concentration, which was in agreement with the results of Farooq et al. (2008) and Chaum et al. (2006).

Foliar application of GB increased total protein concentrations in leaf. Protein accumulation as protective components of the cell can prevent the formation of ice crystals or slow down its formation (Cheng et al., 2004). The accumulation of soluble proteins (similar to the accumulation of soluble carbohydrates) is an adaptation factor reducing the damage caused by low temperatures and thereby increasing the freezing tolerance in plant tissue (Goldsmith, 2009; Guy, 2008). GB protects proteins and cell membranes in abiotic stresses. Sakamoto and Murata (2002) stated that the use of GB stabilized the quaternary structure of the protein complex under osmotic stress. Although GB improved the cold-stress tolerance, perhaps because of its cryoprotective or anti-freeze properties, the mechanism involved is still unknown (Wisniewski et al., 2003). Furthermore, the use of GB protects enzymes such as Rubisco (Makela et al. 1999), which is the most abundant protein of the leaf (Feller et al., 2008). In the present study, application of GB increased the total protein concentration compared to the control, but the highest amount was detected in 2 mM GB treatment. It is consistent with previous

studies that used different concentrations of GB and reported elevated protein concentration (Sakamoto and Murata, 2002).

There are numerous reports of the effect of GB on maintaining relative water content under osmotic stress (Demiral and Türkan, 2004; Farooq et al. 2008; Weibing and Rajaskar, 2018). GB treatment increased the uptake or retention of water in leaves. Maintaining good water status is the most important factor in the usefulness of this amino acid in improving plant condition under osmotic stress. However, Demiral and Türkan (2004) reported a slight decrease in RWC in GB treated strawberry cultivar under normal condition. Increasing RWC by GB treatments was reported by Weibing and Rajaskar (2018) on green bean plant, which confirms findings of this study.

Studies have shown that the cell membrane is the first site of damage due to stress damage (Bajji et al., 2002). Cold stress changes the membrane state from crystal-liquid to solid-gel, thereby disrupts membrane activity (Park et al., 2006). In fact, ionic leakage shows cell membrane injuries (Bajji et al., 2002). Yang and Jian (2010) believed that cold stress increased the electrolyte leakage and the concentration of harmful oxygen compounds. The accumulation of these toxic compounds may lead to the peroxidation of cell membrane lipids and organelles, and may eventually cause physiological disturbance and cold stress injuries in plants. After low temperature damage to the membrane, the cell membrane permeability increases. Any factor that reduces the damage to the membrane and reduces electrolyte leakage can reduce the adverse effects of cold damage (Leul and Zhou, 1999; Bajji et al., 2002; Campos et al., 2003). There have been some reports on the application of GB on reduction of cold damage in plants. For instance, electrolyte leakage was significantly reduced in corn plants treated with GB under cold stress (Chen et al., 2000). The GB seems to protect and stabilize the membrane against intracellular material leaking out of the cell.

GB accumulates in different cellular segments to regulate osmotic pressure, thereby contributing to cell membrane stability.

High correlation between electrolyte leakage and tetrazolium staining was reported earlier by Sarikhani et al. (2014) and Okamoto et al. (2000). In his study, there was a close correlation between the cold resistance measured by electrolyte leakage and the tetrazolium method, which was consistent with the previous results. Results indicated that at lower temperatures, GB in 2 and 1 mM increased membrane viability and protection against cold damage. Tetrazolium staining is previously used to evaluate the cold resistance of various plants such as grapevine (Sarikhani et al., 2014). In this study, T-LT50 value of crown injuries was determined by tetrazolium test, the highest value of which was observed in 2 mM GB treatment with the highest 50% viability obtained at -15.8 °C. The lowest values were also observed in the control treatment at -9.6 °C. Among the concentrations of GB, the highest cold hardiness was observed in 2 mM treated plants followed by 1, 4 and 0.5 mM GB treatments.

The main activity of GB in relation to its effect on reducing the effects of cold stress can be related to its effect on antioxidant system of plant. Although GB by diverse mechanisms such as boosting antioxidant systems, accumulation of compatible solutes, improving preservatives and cell membrane stability can protect the cell and ultimately the plant, antioxidant systems was not investigated in the present study and can be recommended for further investigation.

Conclusion

Low temperature in winter severely restricts the strawberry survival. Application of GB augmented the synthesis of compatible solutes, improved the integrity of cellular membranes and enabled the strawberry plant to tolerate adverse effect of cold temperatures. Based on LT₅₀ calculated from electrolyte leakage as well as tetrazolium staining test, GB especially at 2 mM concentration increased the cold

hardiness in strawberry plant, which was correlated with applied GB concentration, while the lowest amount of cold hardiness was detected in control treatment.

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Conflict of Interest

The authors declared no conflict of interest.

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